# ALKALOIDS OF PETCHIA CEYLANICA

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Abstract—A new alkaloid, petchicine, has been isolated from the stem bark of *Petchia ceylanica* and its structure was established by spectroscopic studies. The <sup>13</sup>C NMR spectra of ajmalicinine, cabucine and fluorocarpamine, which have been isolated from the stem bark of *Petchia ceylanica* for the first time, are also reported.

#### INTRODUCTION

Petchia ceylanica is an evergreen herb indigenous to the low lands of Sri Lanka, and a number of new indole alkaloids have been isolated from it [1-3]. We now report the isolation and structural elucidation of a new aspidosperma alkaloid, (-)-petchicine (1). The <sup>1</sup>H NMR assignments have been made with the help of COSY-45, 2D Jresolved and NOESY experiments while <sup>13</sup>C multiplicities were established by DEPT experiments.

#### RESULTS AND DISCUSSION

The total alkaloids were isolated by extraction of the stem bark of Petchia ceylanica with alcohol, evaporation and extraction of the resulting gum into acid. Purification of the alkaloids and repeated column and TLC afforded petchicine (1) as a colourless solid,  $[\alpha]_D = -380^\circ$ . Its UV spectrum showed  $\lambda_{\rm max}$  at 216, 295 and 328 nm characteristic of an anilinoacrylate chromophore [1, 4, 5]. The IR spectrum showed  $v_{\text{max}}$  3500 (OH), 3550 (N-H) and 1680 (C=O) cm<sup>-1</sup>. The mass spectrum displayed the molecular ion peak at m/z 368.1734 consistent with the molecular formula C<sub>21</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub> (calcd 368.1734) indicating 11 double bond equivalents in the molecule. Other prominent peaks occurred at m/z 336.1695 ( $C_{20}H_{20}N_2O_3$ , 336.1685), 309.1613 ( $C_{19}H_{21}N_2O_4$ , 309.1602), 214. 1587  $(C_{13}H_{12}NO_2, 214.1583, a)$  and 154.0861  $(C_8H_{12}NO_2,$ 159.0867, b). Its fragmentation pattern indicated the presence of an aspidosperma skeleton [1, 4, 5]. The <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, 400 MHz) of petchicine (1) exhibited the presence of 24 protons, each of which was identified with the help of COSY-45, NOESY and 2D-J resolved spectra. A 3H doublet at  $\delta$  1.02 was assigned to the C-18 methyl protons ( $J_{18,19} = 6.4$  Hz), its chemical shift being consistent with the presence of a -CH(OH)Me grouping as in epimisiline [1] or minovincine [5, 6]. A downfield quartet at  $\delta 4.12$  integrated for one proton  $(J_{19,18} = 6.5 \text{ Hz})$  and was assigned to H-19. The COSY spectrum established that it was coupled to a doublet at  $\delta$  1.02 (18-CH<sub>3</sub>) [7] (Fig. 1). A one-proton double-doublet at  $\delta$ 2.05 assigned to C-6<sub> $\alpha$ </sub>H ( $J_{6\alpha, 6\beta} = 11.0$  Hz,  $J_{6\alpha, 5\beta} = 7.0$  Hz) was

coupled to a 1H doublet for C-6 $\beta$ H ( $J_{6\beta,6\alpha}$ =11.0 Hz). A multiplet for C-5 $\beta$ H centred at  $\delta$ 3.70 overlapped with other signals but it was recognized from the 2D-J resolved spectrum [8] to be a double-doublet  $(J_{5\beta,5\alpha})$ = 14.6 Hz,  $J_{5\beta,6\alpha}$  = 7.0 Hz). A 3H singlet for -OMe protons resonated at  $\delta$ 3.75. The connections between C-6 $\alpha$ H, C-5 $\alpha$ H and C-5 $\beta$ H could be seen from the corresponding cross-peaks in the COSY-45 spectrum. A doublet for C-15 $\beta$ H at  $\delta$ 3.09 ( $J_{15\beta, 15\alpha} = 14.4$  Hz) was coupled to the double-doublet for C-15 $\alpha$ H at  $\delta$ 2.95 ( $J_{15\alpha, 15\beta}$  = 14.4 Hz,  $J_{15\alpha,14\alpha}$  = 3.7 Hz). A downfield doublet at  $\delta$ 4.11 for C- $14\alpha H (J_{14\alpha, 3\alpha} = 3.7 \text{ Hz})$  showed strong cross-peaks with signals at  $\delta 3.35$  (C-3 $\alpha$ H),  $\delta 2.95$  (C-15 $\alpha$ H) in the COSY-45 spectrum (Fig. 1). The two 1H signals at  $\delta$ 3.92 and 3.42 were assigned to C-21 $\alpha$ H and C-17 $\alpha$ H respectively, the assignments being confirmed from the NOESY spectrum [9] (Fig. 2). Thus C-17 $\alpha$ H ( $\delta$ 3.42) showed NOESY crosspeaks with the double-doublet at  $\delta 2.05$  (C-6 $\alpha$ H) and  $\delta$ 3.02 (C-5 $\alpha$ H) which not only confirmed the assignment to C-21 aH but it also indicated the x-orientation of the C-21 proton. The aromatic region of petchicine (1) of showed two doublet triplets at  $\delta 6.89$ ( $J_{10,11} = J_{10,9} = 7.6 \text{ Hz}$ ,  $J_{10,12} = 1.2 \text{ Hz}$ ) and  $\delta 7.16$  ( $J_{11,10} = J_{11,12} = 7.6 \text{ Hz}$ ,  $J_{11,9} = 1.0 \text{ Hz}$ ), which were assigned to the C-10 and C-11H respectively. Two doublets integrating for one proton each at  $\delta 6.81$  ( $J_{12,11}$ = 7.6 Hz) and 7.51  $(J_{9,10} = 7.6 \text{ Hz})$  were assigned to the C-12 and C-9 protons respectively, the assignments being supported by the presence of appropriate COSY crosspeaks. The configuration at C-19 was determined by Horeau's procedure [1] from which it was concluded that (-)-petchicine has R configuration at C-19. The multiplet at  $\delta 4.11$  (C-14 $\alpha$ H) showed strong interaction in the NOESY spectrum with a double-doublet at  $\delta$ 3.35 (C- $3\beta$ H) and with the doublet at  $\delta$ 2.95 (C-15 $\alpha$ H), thereby establishing the  $\alpha$ -orientation of the C-14 proton and hence the  $\beta$ -orientation of the ether linkage between C-14/C-17.

The <sup>13</sup>C NMR spectrum (BB and DEPT, 100 MHz, CDCl<sub>3</sub>) of 1 showed resonances for eight methine, four methylene, two methyl and seven quaternary carbon atoms. Comparison of the <sup>13</sup>C NMR spectrum of the non-aromatic part of 1 (Table 1) with the corresponding portions of the <sup>13</sup>C NMR spectra of tabersonine [10] and vincadifformine [11] suggested that there was an ethereal

Dedicated to Prof. Gunther Snatzke on his 60th birthday. \*Author to whom correspondence should be addressed.

oxygen linking two of the carbon atoms in the molecule. The absence of a double bond between C-14 and C-15 (found in tabersonine and a number of other aspidosperma alkaloids) and the replacement of the two signals found in vincadifformine for the C-14 and C-17 methylene carbons by signals for two oxymethine carbons at  $\delta$ 80.0 and 79.4 indicated that the centres C-14 and C-17 were the points of attachment of the ethereal oxygen. The methylene signals at  $\delta$ 56.65, 55.57, 42.57 and 32.90 were assigned to C-3, C-5, C-15 and C-6 respectively. The signal for C-21 appeared at  $\delta$ 78.0, its <sup>13</sup>C chemical shift (along with the <sup>1</sup>H chemical shift of C-21H,  $\delta$ 3.92) indicating  $\alpha$ -orientation of H-21 [1]. The signals of the remaining carbon atoms are given in (Table 1). On the basis of these results structure 1 was assigned to petchicine. (-)-Petchicine belongs to the same enantiomeric series as the other laevo-rotatory alkaloids of aspidospermidine type and its biosynthesis can be conceived through the cyclization of a 14,17-dihydroxylated precur-

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Ajmalicinine (2) was first isolated from Cabucala striolata [12] and its structure was established by chemical and spectroscopic studies (UV, IR, MS and <sup>1</sup>H NMR) [12]. The substance has now been isolated by us from the stem bark of Petchia ceylanica as a white amorphous solid. Its identity was confirmed by spectroscopic studies. The <sup>13</sup>C NMR spectrum (100 MHz, CD<sub>3</sub>COCD<sub>3</sub>) of ajmalicinine (2) (Table 1) showed 21 carbon resonances in agreement with the molecular formula C<sub>21</sub>H<sub>26</sub>N<sub>2</sub>O<sub>4</sub>. The multiplicity assignments were made by DEPT experiment [10], which showed the presence of two methyl, four methylene and ten methine carbon atoms. A lowfield signal at  $\delta$ 91.61 was assigned to the carbon atom bearing two oxygen atoms at C-17 whereas C-19 which bears a single oxygen was found to resonate at  $\delta$ 71.99. Other assignments are presented in Table 1.

Cabucine (3) was previously reported from Cabucalamaga cariensis [13]. We have now isolated it from the stem bark of Petchia ceylanica for the first time as an amorphous solid. The <sup>13</sup>C NMR spectrum (BB and DEPT) of

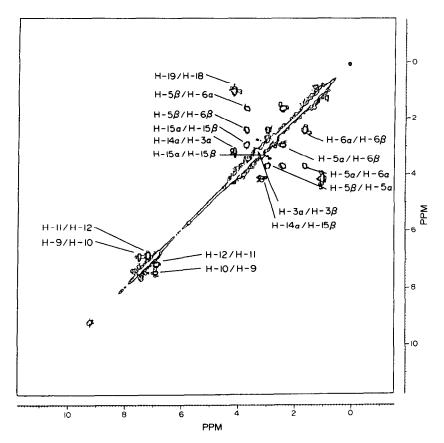


Fig. 1.

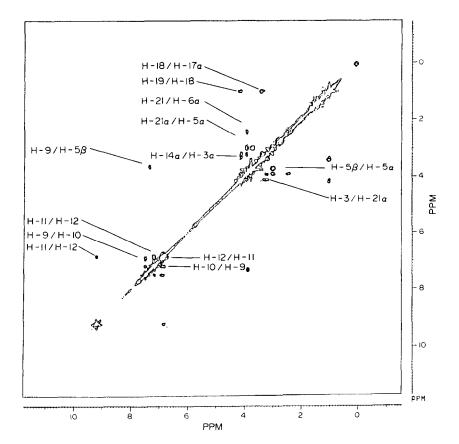


Fig. 2.

C	1	2	3	4
2	160.80	135.85	131.05	W
3	56.65	61.15	60.08	62.11 <sup>b</sup>
5	55.57	53.93	53.16	53.72°
6	32.90a	22.69	21.73	25.61
7	57.21	108.16	107.67	W
8	143.21	128.28	172.63	135.00
9	129.27	111.46	100.40	120.53
10	121.23	119.51	153.98	119.75
11	123.90	121.52	111.41	124.43
12	108.82	111.44	111.03	111.29
13	135.64	135.98	135.78	134.30
14	80.83	35.19	32.77	39.23
15	42.57ª	36.17	30.51	30.60
16	89.50	57.24	106.64	63.32 <sup>t</sup>
17	79.42	91.61	154.00	W
18	14.16	14.65	14.80	12.54
19	64.33	71.99	73.65	121.27
20	47.75	42.77	40.86	137.39
21	78.00	57.59	56.79	55.37

51.65

172.00

50.85

55.89

167.41

51.40

W

Table 1. 13C NMR chemical shifts of compounds 1-4

51.50

OMe

C=O

Ar-OMe

cabucine (100 MHz, CDCl<sub>3</sub>) showed that the non-aromatic region contained four methylene, two methyl and four methine carbon resonances. The characteristic chemical shifts of C-5, C-6 and C-3 carbon atoms at  $\delta$ 53.16, 21.37, and 60.08 respectively are in agreement with presence of a C/D trans ring junction [14, 15]. The position of the methoxy group at C-10 in cabucine was confirmed from the chemical shift values of the aromatic methines of cabucine. It has been reported that if the aromatic ring in the indole moiety is substituted at C-10 position by a methoxy group, the C-9 and C-11 resonances then respectively appear ca 17 and 9.5 ppm upfield from the corresponding unsubstituted compound [16], whereas if the benzene ring of the indole moiety is substituted at C-11 by a methoxy group, the C-10 and C-12 carbons then resonate ca 9.5 and 15 ppm upfield in comparison to the unsubstituted compound [16]. In cabucine (3) the C-9 and C-11 signals were found to resonate at  $\delta$  100.40 and 111.41 while both signals of the unsubstituted counterpart of cabucine, ajmalicine, are known to appear at  $\delta$ 118.1 and  $\delta$ 119 [4]. These results confirmed the substitution of the methoxy group at C-10. The 13C chemical shift values of cabucine are presented in

Fluorocarpamine (4) was also isolated from the stem bark of Petchia ceylanica and its structure confirmed by comparison of spectroscopic data with those reported in the literature [17]. The <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, DEPT) showed eight methine, four methylene and two methyl signals. The two methylene signals at  $\delta$ 53.72 and 55.37 were assigned to the C-5 and C-21 carbon atoms respectively. The C-16 methine carbon bonded to the indole nitrogen resonated at  $\delta$ 63.30 the chemical shifts of other carbons are given in Table 1.

### **EXPERIMENTAL**

The stem bark of Petchia ceylanica were collected from Kakutara district and identified by Prof. S. Balasubramanian (Department of Botany, University of Peredeniya, Peredeniya, Sri Lanka) where a sample specimen is deposited.

Extraction procedure. The ethanolic extract of the air-dried stem bark (50 kg) of Petchia ceylanica collected from the Kekutara district, Sri Lanka, was evapd to a gum (4.1 kg). The total alkaloids were obtained by extraction with CHCl<sub>3</sub> into 5% HCl (101). The ag. layer was then basified to pH 9 with NH<sub>3</sub>, extracted again with CHCl<sub>3</sub> (251) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evapn of this concd CHCl3 extract yielded the crude alkaloids (110 g). The alkaloids were chromatographed on a silica gel column (2 kg) which was successively eluted with mixtures of increasing polarities of petrol, CHCl<sub>3</sub>, Me<sub>2</sub>CO and MeOH. The petrol (40-60°)-CHCl<sub>3</sub> fraction (13:7 to 7:13) was rechromatographed on a column packed with TLC grade silica. Elution was initially with petrol-CHCl<sub>3</sub> (2:1) and then successively with increasing polarities of CHCl<sub>3</sub> and CHCl<sub>3</sub>-MeOH mixtures. A number of alkaloids were isolated from the various fractions.

Isolation of petchicine (1), (-)-Petchicine was isolated from the petrol (40-60°)-CHCl<sub>3</sub> (1:1) eluates obtained from the chromatography column described above, and purified by prep. TLC on silica gel plates using Me<sub>2</sub>CO-petrol (1:2) as eluent as amorphous mass (24.6 mg),  $[\alpha]_D = -380^{\circ}$  (MeOH; c = 0.1 M); UV  $\lambda_{max}^{\text{MeOH}}$  ( $\epsilon$ ): 216 (5.63), 295 (5.03) and 328 (4.76) nm; IR  $\nu_{max}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3500 (OH), 3350 (N-H), 1680 (C=O); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$ : 9.2 (1H, s, N-H), 7.51 (1H, d,  $J_{9,10} = 7.6$  Hz, H-9), 7.16 (1H, dt,  $J_{11,10} = J_{11,12} = 7.6$  Hz,  $J_{11,9} = 1.0$  Hz, H-11), 6.89  $(1H, dt, J_{10,9} = 7.6 \text{ Hz}, J_{10,12} = 1.2 \text{ Hz}, H-10), 6.81 (1H, d, J_{12,11})$ = 7.6 Hz, H-12), 4.21 (1H, q,  $J_{19,18}$  = 6.5 Hz, H-19), 4.11 (1H, d,  $J_{14\alpha, 3\alpha} = J_{14\alpha, 15\alpha} = 3.7 \text{ Hz}, \text{ H-14}\alpha), 3.92 (1\text{H}, \text{s}, \text{H-21}\alpha), 3.75 (3\text{H},$ s, OMe), 3.70 (1H, dd,  $J_{58.5a} = 14.6$  Hz,  $J_{58.6a} = 7.0$  Hz, H-5 $\beta$ ),

<sup>168.69</sup> W=too weak to be measured.

a-c Assignments may be interchangeable.

3.35 (1H, d,  $J_{3\beta,3\alpha} = 14.2$  Hz, H-3 $\beta$ ), 3.09 (1H, d,  $J_{15\beta,15\alpha} = 14.4$  Hz, H-15 $\beta$ ), 3.00 (1H, dd,  $J_{5\alpha,5\beta} = 14.6$  Hz,  $J_{5\alpha,6\beta} = 8.5$  Hz. H-5 $\alpha$ ), 2.05 (1H, dd,  $J_{6\alpha,6\beta} = 12.5$  Hz,  $J_{6\alpha,5\beta} = 7.0$  Hz, H-6 $\alpha$ ), 1.66 (1H, dd,  $J_{6\beta,6\alpha} = 12.5$  Hz,  $J_{6\beta,5\alpha} = 8.5$  Hz, H-6 $\beta$ ); MS m/z (EI, rel. int. %): 368.1734 (100, C<sub>21</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub>), 336 (8), 256 (35), 214 (20), 154 (38), 143 (18), 83 (58);  $^{13}$ C NMR (CDCl<sub>3</sub>, 100 MHz): (Table 1)

Isolation of aimalicinine (2). Aimalicinine was isolated from CHCl<sub>3</sub>-Me<sub>2</sub>CO (2:1) eluates by the column chromatography described above, and purified by prep. TLC on silica gel plates using Me<sub>2</sub>CO-petrol (2:1) as the solvent system. The substance obtained as a white amorphous solid (30 mg), UV  $\lambda_{max}$  (MeOH, e): 225 (4.53), 289 (3.86) and 290 (3.78) nm; IR  $v_{max}^{CHCl_3}$  cm<sup>-1</sup>: 3375 (N-H), 1730 (C=O) and 1625 (C=C); <sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>, 400 MHz)  $\delta$ : 3.25 (dd,  $J_{3\alpha, 14\alpha} = 3.2$  Hz,  $J_{3\alpha, 14\beta} = 11.8$  Hz, H-3 $\alpha$ ),  $2.57 (m, H-5\alpha), 2.63 (m, H-6\alpha), 2.68 (m, H-6\beta), 3.40 (m, H-5\beta), 7.58$ (1H, d,  $J_{9,10} = 7.0$  Hz, H-9), 6.95 (1H, t,  $J_{10,11} = J_{10,9} = 7.0$  Hz, H-10), 7.01 (1H, t,  $J_{11,10} = J_{11,12} = 7.0$  Hz, H-11), 7.25 (1H, d,  $J_{12,11} = 7.0 \text{ Hz}, \text{ H-12}, 2.29 \text{ (1H, } dt, J_{14\alpha,14\beta} = 11.8 \text{ Hz}, J_{14\alpha,3\alpha}$  $=J_{14\alpha,15\alpha}=3.2$  Hz, H-14 $\alpha$ ), 1.29 (1H, q,  $J_{14\beta,14\alpha}=J_{14\beta,15\alpha}$ =11.8 Hz, H-14 $\beta$ ), 2.08 (1H, dd,  $J_{16\alpha,17\alpha} = J_{16\alpha,15\alpha} = 8$  Hz, H-16 $\alpha$ ), 5.09 (1H, d,  $J_{17\alpha,16\alpha} = 8.0$  Hz, H-17 $\alpha$ ), 1.23 (3H, d,  $J_{18,19\beta}$ = 7.0 Hz, H-18), 4.13 (1H, dq,  $J_{19\beta, 18}$  = 7.0 Hz,  $J_{19\beta, 20\beta}$  = 5.0 Hz, H-19 $\beta$ ), 2.07 (1H, m, H-20 $\beta$ ), 2.13 (1H, t, H<sub>21 $\alpha$ , 21 $\beta$ </sub> =  $J_{21<math>\alpha$ , 20 $\beta$ </sub> = 11.0 Hz, H-21 $\alpha$ ), 2.89 (1H, dd,  $J_{21\beta,21\alpha}$ = 11.0 Hz,  $J_{21\beta,20\beta}$ =4.5 Hz, H-21 $\beta$ ) and 3.96 (3H, s, OMe); MS m/z (EI, rel. int.%): 370.1892 (80, C<sub>21</sub>H<sub>26</sub>N<sub>2</sub>O<sub>4</sub>, 396 (100), 184 (42), 170 (45), 169 (30), 156 (43); <sup>13</sup>C NMR (CD<sub>3</sub>COCD<sub>3</sub>, 100 MHz): (Table 1).

Isolation of cabucine (3). Cabucine was isolated from the petrol-CHCl<sub>3</sub> eluates (1:1) from the CC described above and purified as a light yellow-coloured solid mass by prep. TLC on silica gel plates using (2:1) Me<sub>2</sub>CO-petrol. UVλ<sup>McOH</sup><sub>max</sub> (ε): 228 (5.05), 279 (3.94), 295 (3.25) nm; IR  $v_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1668 (C=O), 1620 (C=C);  ${}^{1}H$  NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$ ; 3.38 (1H, bd, H-3 $\alpha$ ), 2.69 (1H, dt,  $J_{5\alpha,5\beta} = 11.3$  Hz,  $J_{5\alpha,6\beta} = 4.2$  Hz, H-5 $\alpha$ ), 2.96 (1H, m, H-6 $\beta$ ), 2.76 (1H, dd,  $J_{6\alpha, 6\beta}$  = 14.0 Hz,  $J_{6\alpha, 5\beta}$  = 4.2 Hz, H-6 $\alpha$ ), 3.09  $(1H, m, H-5\beta)$ , 6.91  $(1H, d, J_{9,11} = 2.4 \text{ Hz}, H-9)$ , 6.78 (1H, dd, $J_{11,12} = 8.7 \text{ Hz}, J_{11,9} = 2.4 \text{ Hz}, H-11$ , 7.18 (1H, d,  $J_{12,11}$ =8.7 Hz, H-12), 3.18 (1H, m, H-14 $\alpha$ ), 1.28 (1H, q,  $J_{14\beta,14\alpha}$  $=J_{14\beta,15\alpha}=11.5$  Hz, H-14 $\beta$ ), 2.42 (1H, m, H-15 $\alpha$ ), 7.52 (1H, d,  $J_{17,15a} = 1.6$  Hz, H-17), 1.18 (3H, d,  $J_{18,19\beta} = 6.6$  Hz, H-18), 4.42 (1H, dq,  $J_{19\beta,18} = 6.6$  Hz,  $J_{19\beta,20\beta} = 5.0$  Hz, H-19 $\beta$ ), 2.15 (1H, m, H-20 $\beta$ ), 2.22 (1H, t,  $J_{21\alpha,21\beta} = J_{21\alpha,20\beta} = 10.8$  Hz, H-21 $\alpha$ ), 2.96  $(1H, m, H-21\beta)$ , 3.73 (3H, s, OMe), 3.84 (3H, s, ArOMe); MS m/z(EI, rel. int. %): 382 (60, C<sub>22</sub>H<sub>26</sub>N<sub>2</sub>O<sub>4</sub>), 367 (24), 325 (18), 295 (40), 200 (70), 199 (38) and 149 (45); 13C NMR (CDCl<sub>3</sub>, 100 MHz) (Table 1).

Isolation of fluorocarpamine (4). Fluorocarpamine was isolated as a yellow coloured substance from the stem bark of Petchia ceylanica. It was isolated from the petrol-CHCl<sub>3</sub> (2:1) eluates and purified by prep. TLC on silica gel plates using CHCl<sub>3</sub>-petrol (2:1). UV  $\lambda_{\max}^{\text{MoOH}}$  ( $\epsilon$ ): 234 (4.05), 255 (3.65), 320

(3.39), 390 (3.13) nm; IR  $\nu_{\max}^{\text{CHC1}_3}$  cm<sup>-1</sup>:1740 (COOMe) and 1668 (C =O); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$ : 7.60 (1H, dd,  $J_{9,10}$  = 7.3 Hz,  $J_{9,11}$  = 1.0 Hz, H-9), 7.51 (1H, t,  $J_{11,10}$  =  $J_{11,12}$  = 7.3 Hz, H-11), 6.91 (1H, t,  $J_{10,11}$  = 7.3 Hz, H-10), 6.70 (1H, dd,  $J_{12,11}$  = 7.3 Hz, H-19, 4.57 (1H, d,  $J_{16,15}$  = 4.3 Hz, H-16 $\alpha$ ), 3.67 (1H, d,  $J_{21\alpha,21\beta}$  = 11.6 Hz, H-21 $\beta$ ), 3.31 (1H, d,  $J_{3\alpha,14\alpha}$  = 4.5 Hz, H-3 $\alpha$ ), 3.05 (1H, dd,  $J_{5\beta,5\alpha}$  = 12.0 Hz,  $J_{5\beta,6\beta}$  = 6.0 Hz, H-5 $\beta$ ), 3.01 (1H, d,  $J_{21\beta,21\alpha}$  = 11.6 Hz, H-21 $\beta$ ), 2.98 (1H, dd,  $J_{5\alpha,5\beta}$  = 12.0 Hz,  $J_{5\alpha,6\alpha}$  = 6.5 Hz, H-5 $\alpha$ ), 2.93 (H, dd,  $J_{6\alpha,6\beta}$  = 12.6 Hz,  $J_{6\alpha,5\beta}$  = 6.0 Hz, H-6 $\alpha$ ); MS: m/z (EI, rel. int. %) 338.1632 (C<sub>20</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub>), 273 (40), 223 (30), 193 (48), 160 (54) and 121 (100); <sup>13</sup>C NMR (CDCl<sub>3</sub>): Table 1.

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